Phytoremediation Potential of Corn and Oat for Increased Levels of Soil Cadmium under Different Irrigation Intervals

A. AZIZIAN[1**], S. AMIN[1*], M. NOSHADI[1*], M. MAFTOUN[2*] and Y. EMAM[3*]

[1]Department of Water Engineering, College of Agriculture, Shiraz University, Shiraz, I.R.Iran
[2] Department of Soil Science, College of Agriculture, Shiraz University, Shiraz, I.R.Iran
[3] Department of Agronomy, College of Agriculture, Shiraz University, Shiraz, I.R.Iran

ABSTRACT-The present pot experiment was undertaken to investigate the phytoremediation potential of corn and oat in soil. Treatments consisted of four cadmium (Cd) levels (0, 5, 10 and 20 mg l^{1}) and three irrigation intervals (1, 3 and 7 days), arranged in the form of a complete randomized design with three replications. Corn and oat were harvested after 75 and 90 days, respectively. Transpiration rates, shoot dry weight and shoot Cd concentration of both plants were measured. Cadmium uptake, Cd bioconcentration factor (BCF), the apparent recovery of Cd and water use efficiency were also calculated. Cadmium had negative, and in some cases stimulating effects on plant growth. Furthermore, the phytoremediation capacities of both plants were higher at the 1 day irrigation frequency. The BCF values for both plants were less than unity, indicating that the phytoremediation potentials of oat and corn were low in this study. Overall, oat was more efficient than corn in phytoremediation of Cd as it accumulated 52, 169 and 132% more Cd than corn at 1, 3 and 7 day irrigation intervals, respectively. On the average, oat also took up soil Cd about 80% more than corn. From the results reported herein, it is recommended to conduct additional experiments with different Cd levels and more irrigation intervals using different types of agronomic and horticultural crops.

Keywords: Cadmium, Corn, Irrigation interval, Oat, Phytoremediation

INTRODUCTION

The development of modern industry and agriculture has led to increased environmental contamination with diverse contaminants among which the increase of heavy metals (e.g Cd) is of prime concern (15 and 7). Although anthropogenic metallic soil pollution is not usually limited to a single metal (22), each element needs to be considered separately because of its unique soil and plant chemistry (4). Cadmium is a non-essential, toxic and dangerous pollutant (9 and 10) which contaminates foods and feeds through the soil-

∗ Former Graduate student, Professor, Associate Professor, Professor, and Professor, respectively
**Corresponding Author

plant-animal chain (13). Therefore, Cd decontamination of soil has a high priority in environmental research. Due to high costs and difficulties in removing toxic metals from soil and water by conventional physical and chemical methods, research is presently focused on low-cost, easy and safe technologies. One such effective and promising procedure is phytoremediation (21), whereby accumulator plants are used to remove toxic metals from soil and water. For phytoremediation to be effective, plants used must have both traits of high biomass and metal accumulation capacity (5). Plants with high metal accumulation capacity usually have a low biomass and are frequently found in sites enriched with heavy metals. In contrast, high biomass plants usually show low metal accumulation capacity (2). Unfortunately, no wild plants or crops have been identified having both capabilities (20). Therefore, phytoremediation research has partly focused on using high biomass plants which can be easily established with common agricultural practice (26), as well as improving them for higher metal uptake by genetic engineering (17). Moreover, various soil factors including water status have a significant impact on phytoremediation. Soil moisture regime influences plant transpiration, which is in turn, directly related to biomass growth. Also, metal absorption by plants is increased as soil water content increases.

Disposal of industrial effluent may be continuous or periodical on nearby lands. In the present study these two situations are simulated by the application of irrigation intervals. The gradual increase in the level of a pollutant through irrigation in the root zone may also abate the problems associated with heavily contaminated soils, such as plant emergence failure, slow growth of plants and depressing the number and diversity of microorganisms, which can negatively affect phytoremediation efficacy (11 and 31).

The present pot study was initiated to evaluate the phytoremediation behavior of corn and oat to gradually elevated soil Cd through different irrigation intervals. These two crops have a valuable role in animal and human nutrition. Furthermore, their shoot abilities for Cd accumulation were reported by several researchers (e.g. 34, 20, 16 and 3). These species of plants can also be established with an agricultural management system and can be easily harvested. In this experiment, shoot dry weight, Cd concentration uptake and several other growth parameters were used to assess the phytremedical ability of these plants.

MATERIALS AND METHODS

Soil source, characteristic and preparation

A bulk sample of the top layer (0-20 cm) of a calcareous silty clay soil (fine, mixed, mesic, Typic calcixerepts) was collected from an uncultivated field at Bajgah Agricultural Experiment Station, 16 km north of Shiraz, Fars province, I. R. of Iran. The soil was air-dried and passed through a 2 mm sieve (some physiochemical properties are shown in Table 1). The soil was mixed uniformly with 150 mg N kg^{-1} as urea, 50 mg P kg^{-1} as KH$_2$PO$_4$, 5 mg Fe kg^{-1} as Fe-EDDHA and 5, 10 and 5 mg Zn, Mn and Cu kg^{-1}, respectively as their sulfate salt. The pots were filled with 6 and 4 kg of treated soil for corn and oat, respectively. The dimensions of pots were 25×30 and 15×20 cm (diameter×depth) for corn and oat respectively. The soil moisture characteristics curve, based on the Van Genuchten Model (32), was

determined using hanging water column and pressure plates apparatus in order to convert the soil moisture to the corresponding matric potential.

Table 1. Physiochemical characteristics of soil used in the study

Soil properties	Amount
Sand (%)	14.1±0.50[†]
Silt (%)	45.5±0.99
Clay (%)	40.4±0.21
Soil texture	Silty clay
Field capacity (dry weight basis, %)	21.64±0.55
Bulk density (Mg/m^3)	1.42±0.11
OM (%)	1.1±0.05
Soil pH (saturated paste)	7.84±0.21
CEC (cmole kg^{-1})	14.1±0.50
EC$_e$ (dS m^{-1})	0.50±0.11
Background Cd (mg kg^{-1})[‡]	0.31±0.09

[†]Mean±S.D. (n=3);
[‡] DTPA extractable

Treatments, plants cultivar and cultivation

The seeds of corn (*Zea mays* L., cv single cross 704) and oat (*Avena sativa* L., cv Moozart) were planted in pots in a greenhouse under natural light with average day and night temperatures of 36° and 11° C, respectively. Each pot was irrigated daily with distilled water to near field capacity by weight, until 15 and 21 days after corn and oat planting, respectively. Then corn and oat were thinned to 3 and 10 plants per pot, respectively, and irrigation was started with water containing different Cd concentrations at 1, 3 and 7 day intervals. Irrigation was done by raising the soil moisture to field capacity by weight. The amounts of applied water were regarded as representative of evapotranspiration, since no water was lost by drainage. Soil moisture was measured by weighing the pots before each irrigation. The Cd levels of irrigation water were 0 (tap water), and 5, 10 and 20 mg/L, obtained by adding CdCl$_2$.H$_2$O to distilled water. These levels were selected to represent reasonable amounts of Cd in raw wastewater (maximum Cd in raw wastewater reported in our country is 0.75 mg/L (Shayegan and Afshari, 2004)). The experimental design was completely randomized, arranged in factorial manner with three replications. In order to measure evaporation, pots without plants were used in each treatment. Irrigation and Cd treatments were continued 75 and 90 days after corn and oat planting, respectively.

Plant harvest and analysis

At harvest, the above-ground portion of the plants was cut at the soil surface. The harvested plant materials were washed with tap and then distilled water, dried to a constant weight at 65°C, and weighed and ground with an electric mill to pass a 40-mesh screen. Representative samples were dry-ashed at 550°C for 4 hrs, extracted with 2M HCl, filtered through Whatman No. 42 filter paper and analyzed for Cd, by an atomic absorption spectrophotometer (model Perkin Elmer 4110).

Data analysis

Cadmium uptake was calculated by multiplying shoot dry weight (DW) and shoot Cd concentration. The total transpiration (T) was obtained by cumulative applied water minus cumulative evaporation. Water use efficiency (WUE) was calculated as the ratio of DW to the total volume of T:

$$WUE = \frac{DW\ (g)}{T\ (l)} \qquad (1)$$

The efficiency of the plants for phytoremediation was evaluated by bioconcentration factor, BCF (the ratio of Cd concentration (*C*) in plant shoot to Cd concentration in the soil (25):

$$BCF = \frac{C_{Shoot}}{C_{Soil}} \qquad (2)$$

Moreover, the apparent Cd recovery was used as the relative efficiency of the two crops for Cd removal from soil as fallow:

$$apparent\ Cd\ recovery\ (\%) = \qquad (3)$$

$$\frac{(Cd\ uptake)_T - (Cd\ uptake)_C}{(Soil\ accumulated\ Cd)_T} \times 100$$

In equation 3, T and C refer to different treatments and controls, respectively.

The data obtained was analyzed using a two-way analysis of variance (ANOVA) for the main and combined effects of irrigation intervals and Cd levels of irrigation water. The differences were statistically significant when the *p*-value was less than 0.05. For comparisons of means, Duncan's multiple range test was used with a 95.0% confidence level.

RESULTS AND DISCUSSION

Soil water depletion, matric potential and Cd concentration

Table 2 summarizes the mean depletion of soil moisture and the produced matric potential that the plants encounter in their growing period. As expected, the maximum soil water content (i.e. minimum soil water depletion) and matric potential occurred at the 1-day irrigation interval. This Table also shows the accumulated soil Cd concentration. The values of this parameter were greater at lower irrigation intervals, because of the larger amount of water application. Despite the lower T rate (Tables 3 and 4), these values were greater for oat than corn which could be attributed to the longer growing period and consequently the need for using more water.

Plant growth

After 50 days, chlorosis appeared on most, especially, old leaves of oat plants, whereas, no such obvious symptoms were observed on corn leaves during the course of the experiment. The ANOVA showed that the irrigation intervals, Cd levels of irrigation water and their interaction effect significantly affected (p≤0.05) the DW and T rate of both plant species in descending order. The only exception was the insignificant interaction effect of irrigation intervals and Cd levels on the T rate of

corn. As expected, the DW and T rate of both plants and their mean values significantly decreased with increasing irrigation frequency (Tables 3 and 4). However, the differences of the DW of oat plant at the 3 and 7 day irrigation interval was not significant when irrigated with 5 mg Cd L^{-1}.

Table 2. Depletion of soil moisture, soil matric potentials and soil Cd concentrations at various irrigation interval and Cd levels of irrigation water

Irrigation interval (days)	Cd level (mg L^{-1})	Soil water depletion (%)		Soil matric potential (bar)		Soil accumulated Cd (mg kg^{-1})	
		Corn	Oat	Corn	Oat	Corn	Oat
1	0	24.9	29.7	-0.40	-0.46	0.00	0.00
	5	23.8	29.9	-0.39	-0.46	11.63	14.56
	10	21.2	27.0	-0.36	-0.43	20.69	26.28
	20	21.5	27.5	-0.36	-0.43	41.98	53.61
3	0	46.9	58.1	-0.77	-1.7	0.00	0.00
	5	46.7	55.2	-0.76	-0.97	7.60	8.97
	10	44.7	59.3	-0.72	-1.10	14.52	19.26
	20	41.3	57.2	-0.65	-1.03	26.84	37.15
7	0	86.9	80.1	-3.91	-2.86	0.00	0.00
	5	86.9	82.9	-3.91	-3.23	6.05	5.78
	10	83.9	84.2	-3.36	-3.44	11.68	11.73
	20	78.2	77.5	-2.66	-2.62	21.78	21.59

The mean DW of corn and oat decreased by increasing the Cd level of irrigation water and dropped significantly, to the minimum amount at the highest Cd level. The plants' T rate was less than that of the control at Cd levels of 10 and 20 mg l^{-1}. On average, the 20 mg L^{-1} Cd caused a reduction of about 17 and 9.5% of the corn and oat T rate, respectively, as compared to the control. In the case of DW, the corresponding values were 21 and 14%. Tables 3 and 4 show that corn had greater potential for biomass production and water consumption. Plant biomass and water uptake contributed markedly to the phytoremediation potential of a given plant species (36 and 21).

Therefore, corn seems to be better for phytoremediation practice than oat. In the present study, a slight increase in DW and T rate was observed with lower doses of Cd levels (5 and 10 mg l^{-1}) especially for oat. Marchiol et al. (23) reported a higher T rate with heavy metals in soil. In contrast, Marchiol et al. (22), Vassilev et al. (33) and Haag-kerwer et al. (12) observed that Cd caused a decline in the T rate of soybean, barley and *Brassica juncea* grown in hydroponic conditions. Plants grown in soil enriched with high levels of heavy metals have lower DW and reduced T rate (19 and 18) while the growth of plants exposed to low Cd levels might be stimulated (19).

Table 3. Shoot dry weight, transpiration rate; shoot Cd concentration and uptake of corn plant at various irrigation intervals and Cd levels of irrigation water

Irrigation interval (days)	Cd level of irrigation water (mg L^{-1})				Mean
	0	5	10	20	
Shoot dry weight (g pot^{-1})					
1	66.52a[*]	64.82a	61.63a	61.35a	63.58A
3	33.40b	30.92b	24.22c	18.02d	26.64B
7	11.50e	11.73e	10.53e	8.64e	10.60C
Mean	37.14A	35.82A	32.13A	29.34B	
Transpiration rate (cm^3 cm^{-2} day^{-1}×10^{-1})					
1	2.37a	2.12b	1.90c	1.93bc	2.08A
3	1.44d	1.36d	1.27de	1.29de	1.34B
7	1.12ef	1.06fg	1.00fg	0.87g	1.01C
Mean	1.64A	1.51B	1.39C	1.36C	
Shoot Cd concentration (mg kg^{-1})					
1	0.24c	1.38abc	1.73ab	2.39a	1.44A
3	0.26c	0.94bc	1.36abc	0.82bc	0.85A
7	0.21c	1.03abc	1.96ab	1.45abc	1.16A
Mean	0.24B	1.12A	1.69A	1.55A	
Cd uptake (μg pot^{-1})					
1	15.37c	86.25b	106.52ab	145.74a	89.22A
3	8.84c	28.92c	32.68c	14.79c	21.31B
7	2.42c	12.17c	21.61c	12.53c	12.17B
Mean	8.87B	43.43A	53.60A	57.68A	

[*]**Means within each parameter followed by the same letter are not statistically different at $p \leq 0.05$. Capital letters are used for the main effects.**

Table 4. Shoot dry weight, transpiration rate, Shoot Cd concentration and uptake of oat plant at various irrigation intervals and Cd levels of irrigation water

Irrigation interval (days)	Cd level of irrigation water (mg L^{-1})				Mean
	0	5	10	20	
Shoot dry weight (g pot^{-1})					
1	16.93a[*]	17.03a	14.84b	14.17b	15.74A
3	10.99c	9.83cd	10.72c	9.44de	10.24B
7	8.18ef	8.65de	8.46ef	7.29f	8.13C
Mean	12.03A	11.82A	11.34A	10.34B	
Transpiration rate (cm^3 cm^{-2} day^{-1}×10^{-1})					
1	1.83a	1.81a	1.61b	1.60b	1.71A
3	1.23c	1.26c	1.24c	1.18c	1.20B
7	0.72d	0.73d	0.72d	0.64d	0.70C
Mean	1.26A	1.27AB	1.19BC	1.14C	
Shoot Cd concentration (mg kg^{-1})					
1	0.47f	6.90c	11.13b	17.63a	9.03A
3	2.17ef	4.03cde	6.13c	10.43b	5.69B
7	1.90ef	2.73def	5.60cd	3.60cdef	3.46C
Mean	1.51D	4.56C	7.62B	10.56A	
Cd uptake (μg pot^{-1})					
1	8.14e	117.87bc	164.69b	251.93a	135.66A
3	25.51e	39.95e	65.61cde	98.42cd	57.37B
7	15.70e	23.18e	47.78de	26.14e	28.2C
Mean	16.45D	60.33C	92.69B	125.48A	

[*]**Means within each parameter followed by the same letter are not statistically different at $p \leq 0.05$. Capital letters are used for the main effects**

Based on the lines of best fit between yield reduction versus Cd concentration, the water and final soil (at the end of the experiment) and shoot Cd concentrations required to reduce shoot dry weight by 10 and 50% were calculated (Table 5). From the results reported in Table 5, corn was more sensitive to Cd under the 3-day irrigation interval. Sameni et al. (29) reported a 50% reduction in the shoot growth of corn at 171 mg Cd kg^{-1}. They also observed that shoot Cd concentration associated with 50% growth suppression was 29 mg kg^{-1} for corn, which was higher than our findings. In our experiment, however, corn was gradually exposed to Cd addition.

In the present study, oat behavior was more complex than corn. According to soil and plant Cd concentrations (Table 5), oat was more tolerant to Cd at the 1-day irrigation frequency. No significant relationship was found between oat growth reduction and shoot Cd concentration under the 7-day irrigation interval.

According to the data in Table 5, oat was more tolerant to Cd than corn under 3 and 7-day irrigation intervals. In general, growth reduction occurred at a higher shoot Cd concentration indicating that oat grew better than corn under Cd stress and might be a preferred candidate for the phytoremediation of Cd.

Table 5. Water, soil and plant tissue Cd concentrations for 10 and 50 % growth reduction of corn and oat at various irrigation intervals

Growth reduction	Irrigation interval (day)					
	1		**3**		**7**	
	Irrigation water Cd concentration (mg L^{-1})					
	Corn	**Oat**	**Corn**	**Oat**	**Corn**	**Oat**
10	21.79	11.70	4.26	14.58	12.80	19.59
50	108.93	58.50	21.31	72.92	27.87	31.16
	Soil Cd concentration (mg kg^{-1})					
10	39.68	31.84	5.09	27.72	12.54	21.42
50	98.41	159.23	25.47	138.58	25.64	33.82
	Shoot Cd concentration (mg kg^{-1})					
10	3.10	12.63	0.40	8.23	1.48	-
50	15.48	32.36	2.02	27.74	7.41	-

Cadmium concentration and uptake

The ANOVA indicated that in corn, shoot Cd concentration was only affected by Cd levels of irrigation water ($p \leq 0.05$), whereas for oat, the effect of both irrigation intervals and Cd levels and also their interactions were significant ($p \leq 0.05$). The same trend was also noted for Cd uptake by either plant. According to Tables 3 and 4, application of Cd in irrigation water resulted in a significant increase in the mean of shoot Cd concentration and uptake by corn plants. However, the mean of Cadmium concentration and uptake for oat markedly increased by increasing Cd levels of irrigation water. An increase in Cd uptake was more pronounced under the lowest irrigation frequency due to higher Cd absorption and more biomass production. At the 1-day irrigation interval, Cd accumulation by corn was 318 and 633% more than the 3 and 7-day irrigation intervals. The corresponding values for oat were 136 and 381%. These results indicate that both plants were effective for removing Cd from soil under optimum soil moisture conditions. Angle et al. (1) observed greater metal uptake and biomass production at higher soil moisture values. The results of our study showed that Cd concentrations in corn shoot are several times less than those of oat. As stated by Jarvis et al. (14), who found that more than

50% of the absorbed Cd was stored in the roots of several plant species, it might be concluded that corn roots have a mechanism which restricts the transfer of Cd through the plant, The biomass production and the accumulation and distribution of metals in the plants' tissue are important aspects of evaluating the role of the plant to remediate contaminated sites (28). Data on shoot dry matter and Cd uptake (Tables 3 and 4) show that despite the lower biomass production, oat accumulated 52, 169 and 132% more Cd than corn at 1, 3 and 7 day irrigation intervals, respectively. On the average, oat took up soil Cd about 80% more than corn.

WUE and BCF

The ANOVA showed that the irrigation interval was the only factor with any significant effect on the WUE of oat (p≤0.05). While in corn, only Cd levels of irrigation water had no significant impact on the WUE. In general, corn had greater WUE than oat, except under the 7-day irrigation frequency. The highest WUE was obtained at 1 and 7-day irrigation intervals for corn and oat, respectively (Table 6). Adequate plant biomass is one of the main effective factors in phytoremediation. Therefore, with a given volume of effluent supply corn may be preferred for phytoremediation under no water stress condition. Both plants had slightly higher WUE when treated with lower doses of Cd (5 or 10 mg l^{-1}) as compared with the control. Marchiol et al. (23) reported lower WUE for crops grown in soil contaminated with heavy metals.

Table 6. WUE and BCF of corn and oat at various irrigation intervals and Cd levels of irrigation water

Plant	Irrigation interval (days)	Cd level of irrigation water (mg L^{-1})				Mean
		0	5	10	20	
Corn				WUE (g l^{-1} water)		
	1	7.62b[*]	8.33ab	8.82a	8.96a	8.43A
	3	6.28c	6.20c	5.16d	4.70d	5.59B
	7	2.79e	3.02e	2.85e	2.68e	2.84C
	Mean	5.57A	5.85A	5.61A	5.45A	
				BCF		
	1	0.787a	0.117b	0.080b	0.057b	0.260A
	3	0.850a	0.120b	0.093b	0.030b	0.273A
	7	0.667a	0.163b	0.160b	0.067b	0.264A
	Mean	0.768A	0.133B	0.111B	0.051B	
Oat				WUE (g L^{-1} water)		
	1	2.70bc	2.75b	2.71bc	2.58bc	2.69B
	3	2.60bc	2.49cd	2.53bcd	2.34d	2.50C
	7	3.32a	3.47a	3.45a	3.34a	3.39A
	Mean	2.87AB	2.90A	2.89A	2.75B	
				BCF		
	1	1.503b	0.463b	0.423b	0.327b	0.679A
	3	6.990a	0.433b	0.317b	0.280b	2.005A
	7	6.130a	0.450b	0.460b	0.167b	1.802A
	Mean	4.874A	0.449B	0.400B	0.258B	

[*]**Means within each parameter followed by the same letter are not statistically different at *p*≤0.05. Capital letters are used for the main effects**

The BCF is an important plant feature in phytoremediation, which implies the uptake of pollutants, their mobilization and storage in the aerial plant parts (23). The BCF values >1 indicate that plants have good ability to take up contaminants from polluted substrate and accumulte them in their above ground tissue. In this study, The BCF of both plants was only affected by Cd treatment ($p \leq 0.05$) and was < 0.5 regardless of Cd rates (Table 6). Marchiol et al. (23) reported the BCF of some heavy metals, including Cd, in various crops. The BCF data of Cd in oat are coherent with those reported by Marchiol et al. (23). In our experiment, oat BCF was greater than that of corn. Therefore, the Cd absorption and translocation ability of oat was greater than corn. The greatest BCF was obtained at the 3-day irrigation interval (Table 6). Our results also showed that corn and oat had greater BCF values under water stress compared to the no soil moisture stress condition. This is mainly due to greater Cd accumulation in the soil resulting from greater amounts of applied water at the 1-day irrigation interval. Robinson et al. (27) also demonstrated a general decrease in the BCF of forest species with increasing soil Cd levels.

Apparent Cd recovery

The Cd recovery percentages are shown in Table 7. According to this Table, at a given Cd level, plants generally recovered more Cd at the 1-day irrigation interval than the other two irrigation regimes. This could be attributed to the higher soil matric potential (Table 2) and consequently greater water and metal uptake and also, higher biomass production (Tables 3 and 4). The amounts of Cd recovered by corn and oat in this study are not effective in phytoremediation *per se*. In the other words, the recovered Cd was very low even under an optimum soil moisture regime, because of the substantial Cd accumulation in the soil. It seems that most of the Cd added to the irrigation water accumulated in the soil and became partially unavailable for plants or especially concentrated in plant roots. From Table 7, it is also clear that rises in Cd levels decreased the percentage of Cd recovery especially for corn. It could thus be realistically hypothesized that corn and oat could perform better in the case of light soil pollution. In general, Cd recovery by oat was greater than corn (Table 7). Based on the results of this study, oat may be selected for further Cd phytoremediation research using actual industrial effluent.

Table 7. Percent of recovered Cd at various irrigation and Cd treatments

Cd level (mg L^{-1})	Irrigation interval (days)	Apparent Cd recovery (%)	
		Corn	Oat
5	1	0.105	0.188
	3	0.044	0.040
	7	0.027	0.032
10	1	0.073	0.149
	3	0.027	0.052
	7	0.027	0.068
20	1	0.027	0.114
	3	0.027	0.049
	7	0.008	0.012

Other agronomic and horticultural crops should also be tested for phytoremediation under gradually increasing levels of soil Cd. In order to artificially enhance Cd uptake, soil amendments may also be used (6 and 8). Finally, the results of this phytoremediation research are recommendable and practical when the experiments are tested in the field under different site-specific and multi-contamination conditions.

CONCLUSION

From a phytoremediation perspective, the results showed that corn and oat responded to water stress by lowering DW, T rate and Cd uptake. They could therefore, perform better under no soil water stress condition, especially under light water and consequently soil Cd pollution. This is because greater metal uptake is observed at higher soil moisture contents and greater metal recovery is obtained with lower Cd levels. Moreover, the results confirm a significant difference between different species for Cd concentrating in shoots: oat seems to be more suitable than corn for Cd phytoremediation.

REFERENCES

1- Angle, J. S., A. J. M. Baker, N. S. Whiting and R. L. Chaney. 2003. Soil moisture effects on uptake of metals by *Thlaspi*, *Alyssum*, and *Berkheya*. Plant Soil 256: 325-332.

2- Baker, A. J. M. and R. R. Brooks. 1989. Terrestrial higher plants which hyperaccumulate metallic elements-a review of their distribution, ecology and phytochemistry. Biorecovery 1: 81-126.

3- Bjerre, G. K. and H. H. Schierup. 1985. Uptake of six heavy metals by oat as influenced by soil type and additions of cadmium, lead, zinc and copper. Plant Soil 88: 57-69.

4- Chaney, R. L., M. Malik, Y. M. Li, S.L. Brown, E. P. Brewer, J. S. Angle and A. J. M. Baker. 1997. Phytoremediation of soil metals. Current Opinion in Biotechnol. 8: 279-284.

5- Chaney, R. L., Y. M. Li, J. S. Angle, E.P. Brewer, J. P. Saunders, and A. J. M. Baker. 1998. Converting metal hyperaccumulator wild plants into commercial phytoremediation systems: Approaches and progress. ***In***: Banũelos, G. S., and N. Terry. **(eds.),** Proceeding of the Fourth International Conference on Biogeochemistry of Trace Elements. Berkeley, C. A. 623-624.

6- Chen, H. and T. Cutright.2001. EDTA and DTPA effects on Cd, Cr, and Ni uptake by helianthus annus. Chemosphere 45: 21-28.

7-Chen H. M., C. R. Zheng, C. T and Z. G. Shen. 2000. Chemical methods and phytoremediation of soil contaminated with heavy metals. Chemosphere 41: 229-234.

8- Chen, Y. X., Q. Lin, Y. M. Luo, Y. F. He, S. J. Zhen, Y. L. Yu, G. M. Tian and M. H. Wong. 2003. The role of citric acid on the phtoremediation of heavy metal contaminated soil. Chemosphere 50: 807-811.

9- Dass, P., S. Samantaray and G. R. Rout. 1997. Studies on cadmium toxicity in plants: A review. Environ. Pollut. 98:29-36.

10- Ekvall, L. and M. Greger. 2003. Effects of environmental biomass-production factors on Cd uptake in two Swedish ecotypes of *Pinus sylvestris*. Environ. Pollut. 21:401-411.

11- Glick B. R. 1995. The enhancement of plant growth by free living bacteria. Can. J. Microbiol. 41: 109-117.

12- Haag-Kerwer, A., H.J. Schäfer, S. Heis, C. Walter and T. Rausch. 1999. Cadmium exposure in *Brassica juncea* causes a decline in transpiration rate and leaf expansion without effect on photosynthesis. J. Exp. Bot. 50: 1827-1835.

13- He, Z. Y., J. C. Li, H.Y. Zhang, and M. Ma. 2005. Different effect of calcium and lanthanum on the expression of phytochelating synthase gene and cadmium absorption in Lactuca sativa. Plant Sci. 168: 309-318.

14- Jarvis, S. C., L. H. P. Jones and M. J. Hopper. 1976. Cadmium uptake from solution by plants and its transport from roots to shoots. Plant Soil 44: 179-191.

15- Jiang, W. H., D. H. Liu and W. Q. Hou. 2001. Hyperacuumulation of cadmium by roots, bulbs and shoots of garlic (*Allium sativum* L.). Bioresour. Technol. 76: 9-13.

16- John, M. K. 1973. Cadmium uptake by eight food crops as influenced by various soil levels of cadmium. Environ. Pollut. 4: 7-15.

17- Kärenlampi, S., H. Schat, J. Vangronsveld, J. A. C. Verkleij, D. Vander Lelie, M. Mergeay and A.I. Tervahauta. 2000. Genetic engineering in the improvement of plants for phytoremediation of metal polluted soils. Environ. Pollut. 107: 225-231.

18- Kastori, R., M. Petrovic and N. Petrovic. 1992. Effect of excess Pb, Cd, Cu and Zn on water relations in sunflower. J. Plant Nut. 15: 2427-2439.

19- Kirkham, M. B. 1978. Water relation of cadmium-treated plants. J. Environ. Qual. 7(3): 334-336.

20- Li, Y. M., R. Chaney, E. Brewer, R. Rosenberg, J. S. Angle, A. J. M. Baker, R. Reeves and J. Nelkin. 2003. Development of a technology for commercial phytoremediation of nickel: economic and technical considerations. Plant Soil 249: 107-115.

21- Licht, L. A. and J. G. Isebrands. 2005. Linking phytoremediation pollutant removal to biomass economic opportunities. Biomass Bioenergy 28: 203-218.

22- Marchiol , L., L. Leita, M. Martin, A. Peressotti and G. Zebri. 1996. Physiological response of tow soybean cultivars to cadmium. J. Environ. Qual. 25: 562-566.

23- Marchiol, L. P. Sacco, S. Assolari and G. Zebri. 2004. Reclamation of polluted soil: phytoremediation potential of crop-related *Brassica* species. Water Air Soil pollut. 158: 345-356.

24- Marchiol, L., S. Assolari, P. Sacco and G. Zebri. 2004. Phytoextraction of heavy metals by canola (*Brassica napus*) and radish (*Raphanus sativus*) grown on multicontaminated soil. Environ. Pollut. 132:21-27.

25- McGrath, S. P. and Zhao F. J. 2003. Phytoextraction of metals and metalloids from contaminated soils. Current Opinion Biotechnol. 14: 277-282.

26- Raskin, I., R. D. Smith and D.E. Salt. 1997. Phytoremediation of metals: using plants to removepollutants from the environment. Current Opinion Biotechnol. 8: 221-226.

27- Robinson, B. H., T. M. Mills, D. Petit, L.E. Fung, S.R. Green and B. E. Clothier. 2000. Natural and induced cadmium-accumulation in poplar and willow: implication for phytoremediation. Plant Soil 227: 301-306.

28- Salt, D. E., M. Blaylock, P. B. A. N. Kumar, V. Dushenkov, B. D. Ensley, I. Chet and I. Raskin. 1998. Phytoremediation: a novel strategy for the removal of toxic metal from the environment using plants. Biotechnol. 13: 468-474.

29- Sameni A. M., M. Maftoun and A. Bassiri. 1987. Response of tomato and sweet corn to different cadmium levels in calcareous soils. J. Hort. Sci. 62(2): 227-232.

30- Shayegan, J. and A. Afshari. 2004. The treatment situation of municipal and industrial wastewater in Iran. J. Water and Wastewater 49: 58-69. (In Farsi)

31- Siciliano, S. D. and J. J. Germida. 1997. Bacteria inoculants of forage grasses enhance degradation of 2-chlorobenzoiz acid in soil. Environ. Toxicol. Chem. 16: 1098-1104.

32- Van Genuchten, M. T. 1980. A closed-form equation for predicting the hydraulic conductivity of unsaturated soils. Soil Sci. Soc. Amer. J. 44: 892-898.

33- Vassilev, A., T. Tsonev, and I. Yordanove. 1998. Physiological response of barly plants (*Hordeum vulgar*) to cadmium contamination in soil during ontogenesis. Environ. Pollut. 103: 287-293.

34- Walker, W. M., J. E. Miller and J. J.Hassett. 1977. Effect of lead and cadmium upon the calcium, magnesium, potassium, and phosphorus concentration in young corn. Soil Sci. 124(3): 145-151.

35- Jiang, W. H., D.H. Liu and W. Q. Hou. 2001. Hyperaccumulation of cadmium by roots, bulbs and shoots of garlic (*Allium sativum* L.) Bioresour. Technol. 76: 9-13.

36- Zhao, F. L., E. Lombi and S. P. Mcgrath. 2003. Assessing the potential for zinc and cadmium phytoremediation with the hyperaccumulator *Thlaspi caerulescens*. Plant Soil 249: 37-43.

ظرفِیت پالایش سبز ذرت و یولاف برای سطوح افزایشی کادمیوم خاک در دورهای مختلف آبیاری

ابوالفضل عزیزیان[1]**،سیف اله امین[1]*،مسعود نوشادی[1]*،منوچهر مفتون[2]* و یحیی امام[3]*

[1]بخش مهندسی آب، دانشکده کشاورزی دانشگاه شیراز، شیراز، جمهوری اسلامی ایران

[2]بخش علوم خاک، دانشکده کشاورزی دانشگاه شیراز، شیراز، جمهوری اسلامی ایران

[3]بخش زراعت و اصلاح نباتات؛ دانشکده کشاورزی دانشگاه شیراز، شیراز، جمهوری اسلامی ایران

چکیده= این آزمایش گلدانی برای بررسی ظرفیت پالایش سبز ذرت و یولاف انجام شد. تیمارها شامل چهار سطح کادمیم (۰، ۵، ۱۰ و $20mg\ L^{-1}$) و سه دور آبیاری (۱، ۳ و ۷ روز) بود که در قالب طرح کاملا تصادفی با سه تکرار اجرا شد. ذرت و یولاف به ترتیب پس از ۷۵ و ۹۰ روز برداشت شد. شدت تعرق، وزن خشک شاخسار و غلظت کادمیم شاخسار هر دو گیاه اندازه گیری شد. جذب کادمیم، ضریب زیستی تجمع (BCF)، بازیافت ظاهری کادمیم و کارایی مصرف آب نیز محاسبه گردید. کادمیم اثر منفی، و در بعضی موارد تحریک کننده، بر رشد داشت. بعلاوه ظرفیت پالایش هر دو گیاه در دور آبیاری ۱ روز بیشتر بود. ضریب زیستی تجمع کادمیم کمتر از یک بود که حاکی از کم بودن ظرفیت پالایش یولاف و ذرت در این آزمایش است. در مجموع یولاف در پالایش کادمیم موثرتر از ذرت بود. بر اساس نتایج بدست آمده آزمایشات بیشتری با تیمارهای مختلف دور آبیاری و سطوح کادمیم با دیگر گیاهان زراعی و باغی توصیه می شود.

واژه های کلیدی: پالایش سبز، دور آبیاری، ذرت، کادمیوم، یولاف